IN THE PRESENCE OF PALLADIUM COMPLEXES

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Although enamines have been well known as an active and important reagent of organic syntheses ( 1 ), their reactions with transition metal complexes have received little attention (2-6). Allylation of active-hydrogen compounds with allylic esters or phenoxides using palladium catalysts was reported to proceed via $\pi-a 11 y l p a l l a d i u m$ complexes $(7,8)$. Tsuji et al. reported briefly the reaction of enamines with $\pi$-allylpalladium chloride (5).

Hence, allylation of enamines with allylic esters or phenoxides is expected to be carried out catalytically. Further, bicycloketones are prepared by the reaction of enamines of c/cloalkanones with 1,4-diacetoxy or 1,4-diphenoxy-2-butene.

A typical procedure is as follows: A mixture of 1 -piperidinocyclohexene ( 0.1 mole), allyl phenoxide ( 0.11 mole ), $\mathrm{Pd}(\mathrm{OAC})_{2}$ ( 0.5 mmole ) and $\mathrm{PPh}_{3}$ ( 2 mmoles ) in abs. benzene ( 50 ml ) was heated at reflux under argon atmosphere ( 6 hr ). After removal of benzene in vacuo, the residue was hydrolyzed by treatment with water ( 100 ml ) at $100^{\circ}(0.5 \mathrm{hr})$ and extracted with ether. The ether extract was washed with aqueous NaOH to remove phenol, concentrated and distiled to give a colorless liquid ( $79-122^{\circ} / 19 \mathrm{~mm}, 12 \mathrm{~g}$ ), a mixture of 2 -allylcyclohexanone ( $70 \%$ ) and 2,6-diallylcyclohexanone ( $13 \%$ ).


Reaction of 1 -piperidinocyclopentene with allyl phenoxide gave similarly a mixture of 2allylcyclopentanone ( $64 \%$ ) and 2,6-diallylcyclopentanone ( $21 \%$ ). Other allylic compounds such as allyl acetate, 1,2 or 3 -methylallyl acetate or phenoxide, etc., were tested successfully to
give the corresponding substituted allyl cycloalkanones. The effective palladium catalysts used were $\mathrm{Pd}(\mathrm{OAC})_{2}-\mathrm{PPh}_{3}, \mathrm{Pd}(\mathrm{AcAc})_{2}-\mathrm{PPh}_{3}$ and $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$. Complexes of other transition metals such as $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}, \mathrm{Pt}(\mathrm{AcAc})_{2}-\mathrm{PPh}_{3}$ and $\mathrm{RuCl}_{2}\left(\mathrm{PPh}_{3}\right)_{3}$ also showed catalytic activity in the order $\mathrm{Pd}->\mathrm{Rh}->\mathrm{Pt}->\mathrm{Ru}$-complex. The fact that the reaction of both crotyl acetate and 3-acetoxy-1butene with 1-piperidinocyclohexene gave the same products, 2-(2-butenyl)cyclohexanone (trans/cis= 4.5), would suggest that these reactions proceed via a common intermediate, a $\pi-1$ methylallylpalladium complex.

If 1,4-diacetoxy or 1,4-diphenoxy-2-butene is used, 4-acetoxy (or phenoxy)-2-butenyl substituted enamine initially formed will again form a $\pi$-allylpalladium complex in which the $\beta^{\prime}$-carbon may attack the $\pi$-allyl group intramolecularly to afford bicycloketones. In anticipation of this hypothesis, synthesis of bicycloketones from enamines of cycloalkanones by single step was carried out.


A mixture of 1,4-diphenoxy-trans-2-butene ( 12.0 g ) (II, $\mathrm{R}=\mathrm{Ph}$ ), $\mathrm{Pd}(\mathrm{OAC})_{2}$ (1 mmole), $\mathrm{PPh}_{3}$ (4 mmoles) and l-piperidinocyclopentene ( 7.7 g ) ( $\mathrm{I}, \mathrm{n}=0$ ) in abs. benzene was heated under argon atmosphere ( 3 hr ). After the usual work-up, the products were submitted to silica gel chromatography. Distillation of the eluate gave bicyclo[4.2.1]non-3-en-9-one (III, $n=0$ ) ( $1.9 \mathrm{~g}, 28 \%$, bp $72-73^{\circ} / 4 \mathrm{~mm}, \mathrm{mp} 49-50^{\circ}$ ), m/e $=136$, $\mathrm{ir} ; 3025 \mathrm{~cm}^{-1}(=\mathrm{CH}), 1740(\mathrm{C}=0), \mathrm{nmr} ; \delta 5.58$ (Ha), 2.93 $(\mathrm{HC}), 2.15(\mathrm{Hb}), 2.1-1.5$ (other ring protons). 2,4-Dinitrophenylhydrazone, mp 182-183. Hydrogenation over palladium sponge gave bicyclo[4.2.1]nonan-9-one, mp 109-110 (9). The results using other catalysts such as $\mathrm{Pd}(\mathrm{AcAc})_{2}-\mathrm{PPh}_{3}$ and $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}$ - Ph 0 Na were virtually identical with that described above.

Reaction of 1-pyrrolidinocyclohexene with 1,4-diacetoxy-2-butene (II, $R=0 A C$ ) afforded a mixture of three products, bicyclo[4.3.1]dec-3-en-10-one (III, $n=1$ ), 6-exo-vinyl-bicyclo[3.2.1]-octan-8-one (IV, $n=1$ ) and 6-endo-vinyl-bicyclo[3.2.1]octan-8-one (IV, $n=1$ ). Each compound was separated by preparative vpc , and the structure was assigned on the basis of the following data.

Bicyclo[4.3.1]dec-3-en-10-one, mp 58-61 ${ }^{\circ}$, $\mathrm{m} / \mathrm{e}=150$, ir; $3015 \mathrm{~cm}^{-1}(=\mathrm{CH}), 1700(\mathrm{C}=0)$, nmr ; $\boldsymbol{\delta} 5.75(\mathrm{Ha}), 2.63(\mathrm{Hc}), 2.28(\mathrm{Hb}), 2.1-1.3$ (other ring protons). 2,4-Dinitrophenylhydrazone, mp 174-175 ${ }^{\circ}$.

The configurational assignment (exo- or endo-vinyl) of compounds (IV) was based on the chemical shift of the Ha proton. In 2- substituted norbornanes, the absorption of the endo proton usually appears upfield than that of the corresponding exo proton (10, 11).

6-exo-Vinyl-bicyclo[3.2.1]octan-8-one, ir; $1745 \mathrm{~cm}^{-1}(\mathrm{C}=0)$, 994 and 915 (terminal vinyl), $\mathrm{nmr} ; \mathcal{5} 5.18(\mathrm{Hb}), 4.89(\mathrm{HC}), 4.83(\mathrm{Hd}), 2.62(\mathrm{Ha}), 2.3-1.5$ (other ring protons). The seven lines ( Hb ) collapsed to four lines on irradiation at $\delta 2.62$ ( Ha ). 2,4-Dinitrophenylhydrazone, mp 127130 ${ }^{\circ}$. 6-endo-Vinyl-bicyclo[3.2.1]octan-8-one, $\mathrm{ir} ; 1750 \mathrm{~cm}^{-1}(\mathrm{C}=0)$, 993 and 915 (terminal vinyl), $\mathrm{nmr} ; \delta 6.02(\mathrm{Hb}), 5.15(\mathrm{Hd}), 5.13(\mathrm{Hc}), 2.84(\mathrm{Ha}), 2.4-1.4$ (other ring protons). The seven lines (Hb) collapsed to four lines on irradiation at $\delta 2.84$ (Ha). 2,4-Dinitrophenylhydrazone, mp 113$115^{\circ}$.

Similar reactions were performed on 1-pyrrolidinocycloheptene and 1-pyrrolidinocyclooctene, and the results are surmarized in Table 1.

Table 1. Bicycloketones from Cycloalkanone Enamines

| (I) | R of (II) | $n$ | (III) (\%) | (IV) (endo/exo) | (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1-Piperidinocyclopentene | $\mathrm{COCH}_{3}$ | 0 | 17 | - |  |
|  | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 0 | 28 | - |  |
| 1-Pyrrolidinocyclohexene | $\mathrm{COCH}_{3}$ | 1 | 30 | 34 (0.41) |  |
| 1-Pyrrolidinocycloheptene | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 2 | 28 | 39 (2.21) |  |
| 1-Pyrrolidinocyclooctene | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 3 | 12 | 28 (0.31) |  |
| A mixture of (I) ( 0.05 mole ), (II) ( 0.05 mole ), $\mathrm{Pd}(\mathrm{OAC})_{2}$ ( 1 mmole ) and $\mathrm{PPh}_{3}$ |  |  |  |  |  |

These reactions would be rationalized as shown in Scheme 1. Enamines (I) react with the $\pi$-allylpalladium complexe derived from (II) to give 4-phenoxy(or acetoxy)-2-butenyl substituted enamines (V). The enamines (V) also form $\pi$-allylpalladium complexes (VI), in which the $\boldsymbol{\beta}^{\prime}$-carbon attacks the $\pi-a l l y l$ group intramolecularly either path a or $b$ followed by hydrolysis to give bicycloketones (III) or (IV). In fact, when reaction of l-piperidinocyclopentene with 1,4-di-acetoxy-2-butene was stopped in the early stage, formation of (III, $n=0$ ) was negligible, and was
obtained a hydrolysis product of an intermediate (V) ( $n=0, R=\mathrm{COCH}_{3}$ ), 2-(4-acetoxy-trans-2butenyl)cyclopentanone. Ir; 1720 and $1740 \mathrm{sh}(\mathrm{C}=0$ ). 2,4-Dinitrophenylhydrazone, mp 98-100 , ir; $1730 \mathrm{~cm}^{-1}(\mathrm{C}=0)$.

Scheme 1.
(I)


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