REACTION OF CYCLOALKANONE ENAMINES WITH ALLYLIC COMPOUNDS

IN THE PRESENCE OF PALLADIUM COMPLEXES

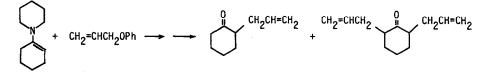
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(Received in Japan 11 November 1972; received in UK for publication 8 December 1972)

Although enamines have been well known as an active and important reagent of organic syntheses (1), their reactions with transition metal complexes have received little attention (2-6). Allylation of active-hydrogen compounds with allylic esters or phenoxides using palladium catalysts was reported to proceed <u>via</u> π -allylpalladium complexes (7,8). Tsuji et al. reported briefly the reaction of enamines with π -allylpalladium chloride (5).

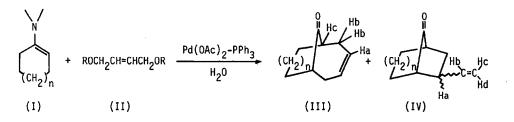
Hence, allylation of enamines with allylic esters or phenoxides is expected to be carried out catalytically. Further, bicycloketones are prepared by the reaction of enamines of c/cloalkanones with 1,4-diacetoxy or 1,4-diphenoxy-2-butene.

A typical procedure is as follows: A mixture of 1-piperidinocyclohexene (0.1 mole), allyl phenoxide (0.11 mole), $Pd(OAc)_2$ (0.5 mmole) and PPh_3 (2 mmoles) in abs. benzene (50 ml) was heated at reflux under argon atmosphere (6 hr). After removal of benzene <u>in vacuo</u>, the residue was hydrolyzed by treatment with water (100 ml) at $100^{\circ}(0.5 \text{ hr})$ and extracted with ether. The ether extract was washed with aqueous NaOH to remove phenol, concentrated and distilled to give a colorless liquid (79-122°/19 mm, 12 g), a mixture of 2-allylcyclohexanone (70 %) and 2,6-di-allylcyclohexanone (13 %).



Reaction of 1-piperidinocyclopentene with allyl phenoxide gave similarly a mixture of 2allylcyclopentanone (64 %) and 2,6-diallylcyclopentanone (21 %). Other allylic compounds such as allyl acetate, 1,2 or 3-methylallyl acetate or phenoxide, etc., were tested successfully to give the corresponding substituted allyl cycloalkanones. The effective palladium catalysts used were $Pd(OAc)_2-PPh_3$, $Pd(AcAc)_2-PPh_3$ and $Pd(PPh_3)_4$. Complexes of other transition metals such as $RhCl(PPh_3)_3$, $Pt(AcAc)_2-PPh_3$ and $RuCl_2(PPh_3)_3$ also showed catalytic activity in the order Pd->Rh->Pt->Ru-complex. The fact that the reaction of both crotyl acetate and 3-acetoxy-1-butene with 1-piperidinocyclohexene gave the same products, 2-(2-butenyl)cyclohexanone (trans/cis= 4.5), would suggest that these reactions proceed <u>via</u> a common intermediate, a π -1-methylallylpalladium complex.

If 1,4-diacetoxy or 1,4-diphenoxy-2-butene is used, 4-acetoxy (or phenoxy)-2-butenyl substituted enamine initially formed will again form a π -allylpalladium complex in which the β -carbon may attack the π -allyl group intramolecularly to afford bicycloketones. In anticipation of this hypothesis, synthesis of bicycloketones from enamines of cycloalkanones by single step was carried out.



A mixture of 1,4-diphenoxy-trans-2-butene (12.0 g) (II, R=Ph), $Pd(OAc)_2$ (1 mmole), PPh_3 (4 mmoles) and 1-piperidinocyclopentene (7.7 g) (I, n=0) in abs. benzene was heated under argon atmosphere (3 hr). After the usual work-up, the products were submitted to silica gel chromato-graphy. Distillation of the eluate gave bicyclo[4.2.1]non-3-en-9-one (III, n=0) (1.9 g, 28 %, bp 72-73°/4 mm, mp 49-50°), m/e= 136, ir; 3025 cm^{-1} (=CH), 1740 (C=0), nmr; \$ 5.58 (Ha), 2.93 (Hc), 2.15 (Hb), 2.1-1.5 (other ring protons). 2,4-Dinitrophenylhydrazone, mp 182-183°. Hydrogenation over palladium sponge gave bicyclo[4.2.1]nonan-9-one, mp 109-110° (9). The results using other catalysts such as Pd(AcAc)_2-PPh_3 and PdCl_2(PPh_3)_2-PhONa were virtually identical with that described above.

Reaction of 1-pyrrolidinocyclohexene with 1,4-diacetoxy-2-butene (II, R= OAc) afforded a mixture of three products, bicyclo[4.3.1]dec-3-en-10-one (III, n=1), 6-exo-viny1-bicyclo[3.2.1]octan-8-one (IV, n=1) and 6-endo-viny1-bicyclo[3.2.1]octan-8-one (IV, n=1). Each compound was separated by preparative vpc, and the structure was assigned on the basis of the following data. No. 2

Bicyclo[4.3.1]dec-3-en-10-one, mp 58-61°, m/e= 150, ir; 3015 cm⁻¹(=CH), 1700 (C=O), nmr; δ 5.75 (Ha), 2.63 (Hc), 2.28 (Hb), 2.1-1.3 (other ring protons). 2,4-Dinitrophenylhydrazone, mp 174-175°.

The configurational assignment (exo- or endo-vinyl) of compounds (IV) was based on the chemical shift of the Ha proton. In 2- substituted norbornanes, the absorption of the endo proton usually appears upfield than that of the corresponding exo proton (10, 11).

6-exo-Vinyl-bicyclo[3.2.1]octan-8-one, ir; 1745 cm⁻¹(C=O), 994 and 915 (terminal vinyl), nmr; δ 5.18 (Hb), 4.89 (Hc), 4.83 (Hd), 2.62 (Ha), 2.3-1.5 (other ring protons). The seven lines (Hb) collapsed to four lines on irradiation at δ 2.62 (Ha). 2,4-Dinitrophenylhydrazone, mp 127-130°. 6-endo-Vinyl-bicyclo[3.2.1]octan-8-one, ir; 1750 cm⁻¹(C=O), 993 and 915 (terminal vinyl), nmr; δ 6.02 (Hb), 5.15 (Hd), 5.13 (Hc), 2.84 (Ha), 2.4-1.4 (other ring protons). The seven lines (Hb) collapsed to four lines on irradiation at δ 2.84 (Ha). 2,4-Dinitrophenylhydrazone, mp 113-115°.

Similar reactions were performed on 1-pyrrolidinocycloheptene and 1-pyrrolidinocyclooctene, and the results are summarized in Table 1.

| (I) | R of (II) | n | (III) (%) | (IV) (endo/exo) (%) |
|---------------------------|-------------------------------|---|-----------|---------------------|
| 1-Piperidinocyclopentene | соснз | 0 | 17 | |
| | с ₆ н ₅ | 0 | 28 | |
| l-Pyrrolidinocyclohexene | соснз | ۱ | 30 | 34 (0.41) |
| 1-Pyrrolidinocycloheptene | с _б н ₅ | 2 | 28 | 39 (2.21) |
| 1-Pyrrolidinocyclooctene | с ₆ н ₅ | 3 | 12 | 28 (0.31) |

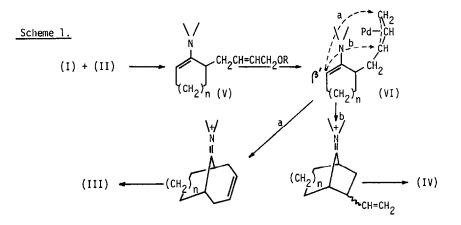
Table 1. Bicycloketones from Cycloalkanone Enamines

A mixture of (I) (0.05 mole), (II) (0.05 mole), Pd(OAc)₂ (1 mmole) and PPh₃

(4 mmoles) in abs. benzene (100 ml) was refluxed for 3 hr under argon.

These reactions would be rationalized as shown in Scheme 1. Enamines (I) react with the π -allylpalladium complexe derived from (II) to give 4-phenoxy(or acetoxy)-2-butenyl substituted enamines (V). The enamines (V) also form π -allylpalladium complexes (VI), in which the e'-carbon attacks the π -allyl group intramolecularly either path a or b followed by hydrolysis to give bicycloketones (III) or (IV). In fact, when reaction of 1-piperidinocyclopentene with 1,4-di-acetoxy-2-butene was stopped in the early stage, formation of (III, n=0) was negligible, and was

obtained a hydrolysis product of an intermediate (V) (n=0, R=COCH₃), 2-(4-acetoxy-trans-2-butenyl)cyclopentanone. Ir; 1720 and 1740 sh (C=0). 2,4-Dinitrophenylhydrazone, mp 98-100°, ir; 1730 cm⁻¹(C=0).



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